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Near-Unity Efficiency Energy Transfer from Colloidal Semiconductor Quantum Wells of CdSe/CdS Nanoplatelets to A Monolayer of MoS₂

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Abstract

A hybrid structure of the quasi-2D colloidal semiconductor quantum wells assembled with a single-layer of 2D transition metal dichalcogenides offers the possibility of highly strong dipole-to-dipole coupling, which may enable extraordinary levels of efficiency in Förster resonance energy transfer (FRET). Here, we show ultra-high efficiency FRET from the ensemble thin films of CdSe/CdS nanoplatelets (NPLs) to a MoS₂ monolayer. From time-resolved fluorescence spectroscopy, we observed the suppression of the photoluminescence of the NPLs corresponding to the total rate of energy transfer from ~ 0.4 to 268 ns^{-1} . Using an Al₂O₃ separating layer between CdSe/CdS and MoS₂ with thickness tuned from 5 to 1 nm, we found that FRET takes place 7 to 88-fold faster than the Auger recombination in CdSe-based NPLs. Our measurements reveal that the FRET rate scales down with d^{-2} for the donor of CdSe/CdS NPLs and the acceptor of MoS₂ monolayer, d being the center-to-center distance between this FRET pair. A full electromagnetic model explains the behavior of this d^{-2} system. This scaling arises from the delocalization of the dipole fields in the ensemble thin film of the NPLs and full distribution of the electric field across the layer of MoS₂. This d^{-2} dependency results in an extraordinary long Förster radius of ~ 33 nm.

KEYWORDS: *nonradiative energy transfer, FRET, colloidal nanoplatelets, molybdenum disulphide, distance dependency, Förster radius, Auger recombination*

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3 Two-dimensional (2D) materials, also known as van der Waals materials, show diverse electronic
4 structures and properties. Among them are examples of transition metal dichalcogenide (TMDC)
5 semiconductors (*e.g.*, molybdenum disulphide, MoS₂),¹ semimetals (*e.g.*, graphene)² and
6 insulators (*e.g.*, hexagonal boron nitride, hBN).¹ Specifically, semiconducting TMDC 2D crystals
7 including MoS₂ and WSe₂ exhibit very strong light-matter interaction.³ For instance, MoS₂
8 monolayer can absorb as much as 10% of the normal incidence light at its excitonic peaks (615
9 and 660 nm).^{1,2} This strong optical interaction has attracted great attention for applications in
10 optoelectronics,³⁻⁵ nanophotonics^{2,5} and sensing.⁶⁻⁸ More recently, another important class of
11 semiconductors with their quasi-2D structure, colloidal quantum wells (CQWs), the so-called
12 nanoplatelets (NPLs), which are most commonly synthesized in the form of CdX (where X
13 represents S, Se and Te), have attracted interest for the ability to precisely control their vertical
14 thickness.⁹⁻¹¹ Owing to the strong quantum confinement in the vertical direction, these
15 semiconductor NPLs demonstrate attractive optical properties including giant oscillator strength,
16 and narrow photoluminescence (PL) emission (purely homogenous broadening).¹¹⁻¹⁴ These
17 interesting properties of the NPLs make them a highly promising material platform for efficient
18 light-emitting diodes (LEDs),¹⁵ lasers,^{11,12} and photovoltaic devices.^{16,17}

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40 Nonradiative energy transfer (NRET) is an electromagnetic process of the energy transfer from a
41 donor to an acceptor by near-field dipole-dipole coupling. One type of NRET is Förster resonance
42 energy transfer (FRET),^{18,19} where the rate of the excitation energy transfer essentially depends on
43 the distance between the donor and the acceptor as well as the spectral overlap between the PL
44 spectrum of the donor and the optical absorption spectrum of the acceptor, the absorption cross-
45 section of the acceptor, PL quantum yield (QY) of the donor, and orientation of the dipoles between
46 the donor and the acceptor.¹⁸⁻²⁰ FRET process has been shown to be effectively enhance the
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3 performance of the light-generation, -harvesting and -sensing functions in specially designed
4 hybrid structures. To date, numerous FRET studies have been conducted using colloidal
5 semiconductor nanocrystals with different degrees of quantum confinement (quantum dots (QDs),
6 nanorods (NRs), NPLs).²⁰ For example, FRET has been investigated in a hybrid structure of two
7 types of NPLs, each with a different vertical thickness.^{21,22} Thanks to their large absorption cross-
8 section, NPLs emerged as a highly promising candidate for absorbing photons and photogenerating
9 excitons potentially to be transferred *via* FRET. To this end, graphene,^{23,24} graphene oxide^{25,26} and
10 carbon nanotubes²⁷ have been used as a sinking medium of the excitons. Especially, TMDC
11 semiconductors including MoS₂^{28,29} and MoSe₂³⁰ have exhibited high capability of exciton sinking
12 from neighboring nanoemitters. For example, the FRET pairs of CdSe/CdS QD-donors with
13 MoSe₂-acceptors³⁰ and of CdSe/CdS QD-donors with graphene-acceptors,³¹ which are shown to
14 exhibit the distance scaling of d^{-4} , have previously been reported. However, to the best of our
15 knowledge, FRET has not been explored in the hybrid structure of the 2D NPLs and TMDC,
16 although such a combination is very favorable because of their intrinsic geometrical similarity.
17 Owing to their atomically flat topologies, the layer-to-layer dipole-dipole interaction between
18 NPLs and TMDC films would be expected to achieve ultraefficient FRET compared to any other
19 possible combinations of nanocrystal semiconductors and 2D materials.

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22 In the present study, we report FRET from solid assembled thin films of colloidal CdSe/CdS
23 core/crown NPLs serving as the donor to a single layer of MoS₂ serving as the acceptor. Here, we
24 systematically study the FRET process and surprisingly find the rate of the FRET decays as $1/d^2$,
25 where d is the distance between the donor NPL and acceptor TMDC 2D-structures. Although the
26 decay rate is expected to scale with $1/d^4$, we reveal that the strong field screening and the dipole
27 electric field delocalization at the donor play major role for modifying the decay kinetics. To study
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3 systematically FRET as function of the distance between the NPL nanoemitters and MoS₂
4 monolayer, we deposited thin films of dielectric spacers of alumina (Al₂O₃) with tuned thicknesses
5 between the donor and the acceptor precisely controlled using atomic layer deposition (ALD) and
6 employed the time-resolved fluorescence (TRF) spectroscopy to investigate the resulting
7 luminescence decays of the NPLs. According to the measurements, the PL curves of the NPLs
8 were significantly accelerated with decreasing thickness of Al₂O₃, which convincingly indicates
9 the existence of the energy transfer. Subsequently, we found that by decreasing the spacer
10 thickness to 1 nm, we achieved near-unity energy transfer efficiency of 99.88% with the ultrafast
11 FRET lifetime of 3.73 ps at room temperature. Additionally, we support our findings with a full
12 numerical electromagnetic model of the hybrid structure by simulating the dipole interaction
13 within the model structure. We believe that this proposed system can offer interesting opportunities
14 to achieve highly efficient photovoltaics devices, ultralow-threshold colloidal lasers and
15 photocatalytics.

32 33 34 35 **Results and Discussion**

36 The synthesized 4 monolayer (ML) CdSe/CdS core/crown nanoplatelets (see Supporting
37 Information for details) were transferred onto the pre-cleaned substrates *via* spin-coating. To study
38 the distance dependency of energy transfer between the donor and the acceptor, thin films of Al₂O₃
39 were deposited on the top of a monolayer of MoS₂ *via* ALD technique. We measured the thickness
40 of Al₂O₃ layer using ellipsometry technique (see Supporting Information, Figure S3). The
41 thickness of the MoS₂ films was attained *via* atomic force microscopy (AFM) and we further
42 performed Raman spectroscopy to verify the monolayer formation of MoS₂ flakes. (see Supporting
43 Information, Figure S2). Figure 1a shows the schematic representation of the fabricated hybrid
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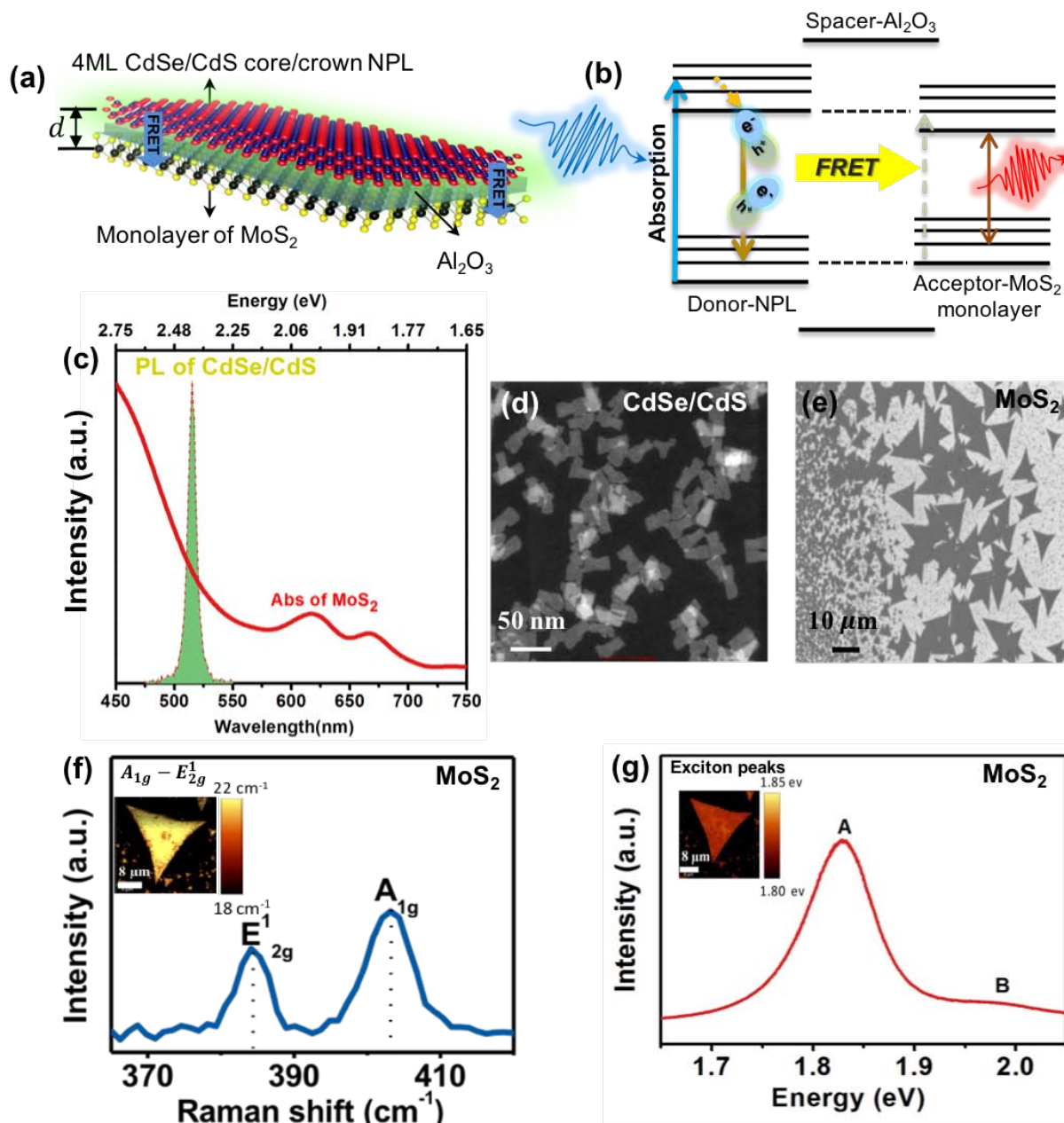


Figure 1. (a) Graphical illustration of the hybrid assembled films of NPLs-Al₂O₃-MoS₂ (donor- spacer-acceptor) system. (b) Schematic representation of the FRET process where the energy band diagram indicates the transfer of excitons. (c) Normalized photoluminescence of the 4ML CdSe/CdS core/crown NPLs (donor) and UV-vis absorption spectrum of the MoS₂ triangular monolayer flakes (acceptor). (d) High-angle annular dark field transmission microscopy image of the NPLs. (e) Scanning electron microscopy image of the MoS₂ film. (f) Raman spectrum of the MoS₂ film. The inset shows the $A_{1g} - E_{2g}^1$ central map of Raman spectra. (g) PL spectrum of the MoS₂ film. The inset exhibits the spatial map of exciton peak around 1.83 eV.

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3 structure of the 4 monolayer CdSe/CdS core/crown NPLs and MoS₂ monolayer, which are
4 separated from each other with the dielectric thin films of Al₂O₃ having thickness tuned from 1 to
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6 40 nm. Here in, d is the center-to-center distance between the donor and the acceptor dipoles (
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8 where $d = d_L + t_{NPL}/2 + t_{Al_2O_3}$ where the d_L is the length of oleic acid (calculated in the range
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10 of 1.95-1.99 nm),^{32,33} t_{NPL} is the thickness of NPLs (taken as 1.2 nm for 4 ML CdSe core
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12 NPLs)^{32,33} and $t_{Al_2O_3}$ is the thickness of Al₂O₃). Figure 1b schematically illustrates the FRET
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14 process between the donor and the acceptor where the excited donor (D^*) relaxes to its ground
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16 state (D) while the excitation energy is transferred to the acceptor, which it is in the ground state
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18 (A) before the energy transfer is occurred. ($D^* + A \xrightarrow{FRET} D + A^*$). In our particular study, D is the
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20 CdSe/CdS NPL with 4 monolayers, while A is the single MoS₂ monolayer. Figure 1c shows the
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22 spectral overlap between the PL spectrum of the donor and the UV-vis absorption spectrum of the
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24 acceptor. PL spectrum of the NPLs peaks at 518 nm with a full-width at half maximum (FWHM)
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26 of 8 nm and the two peaks in the absorption spectrum of MoS₂ around 1.83 and 2.0 eV,
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28 corresponding to A and B excitonic peaks, respectively. From the high-angle annular dark field
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30 transmission electron microscopy (HAADF-TEM) images of the NPLs (Figure 1d), the average
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32 length and width are calculated to be 34.37 ± 2.21 nm and 15.67 ± 1.29 nm, respectively. Figure
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34 1e depicts the scanning electron microscopy (SEM) image of the isolated MoS₂ triangular flakes
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36 where the surface coverage of these islands reaches up to 80%. The Raman shifts and the PL
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38 spectrum of the MoS₂ flakes are presented in Figure 1f and 1g, respectively. The inset of Figure
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40 1f exhibits the $A_{1g} - E'_{2g}$ Raman mode difference map, which is in the range of 18 to 22 cm⁻¹,
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42 typical for MoS₂ flakes.^{34,35} Furthermore, the PL spectrum (Figure 1g) and A exciton peak center
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44 map (the inset of Figure 1g) exhibit the A and B exciton peaks and a bandgap of ~1.83 eV,
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46 respectively, confirming the growth of MoS₂ monolayer.³⁶ In addition, we further determined the
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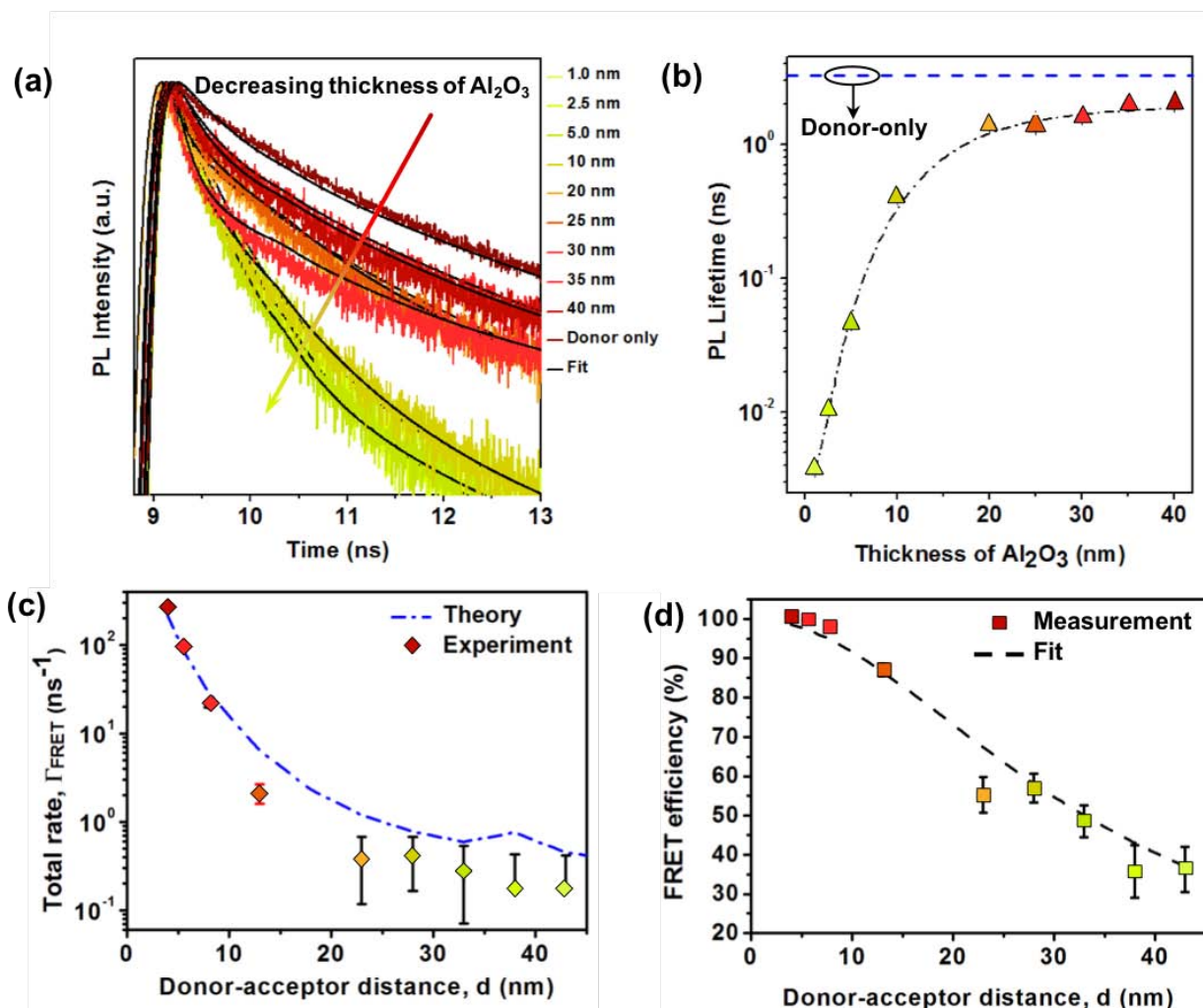


Figure 2. (a) Photoluminescence decay curves of the solid films of the NPLs-MoS₂ assembly, collected at the donor PL emission peak with varying thicknesses of the Al₂O₃ separation layer. The solid black lines represent the fits of the curves. (b) Evolution of amplitude-averaged lifetime of the solid film samples at PL emission peak wavelength of the donor as a function of the spacer layer thickness. The blue dashed-line indicates the only donor PL lifetime. (c) Total rate of FRET as function of the Al₂O₃ thin-film thickness (separation thickness). The blue dot-dashed line is electromagnetic behavior of the system (COMSOL electromagnetic numerical simulation). (d) Efficiency of FRET as a function of the center-to-center distance between the donor and the acceptor. The dashed line is the numerical fitting.

thickness of MoS₂ flakes *via* atomic force microscopy (Supporting Information, Figure S2), which corroborates the formation of the single layer flakes.

TRF decays of the donor PL bandedge emission (at 518 nm) are shown in Figure 2c for varying separating thickness of the dielectric layer. To verify that the NPLs are distributed homogeneously on the films, we performed the TRF measurements at the different positions on the samples and observed that the PL decay curves are very similar in each position (see Supporting Information,

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3 Figure S4 and Tables S1-S10). Here, the donor-only sample, in which the NPLs were deposited
4 alone on a pre-cleaned Al₂O₃-SiO₂-Si substrate where the thickness of Al₂O₃ is 5 nm, acts as the
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6 reference sample.
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10 As can be seen in Figure 2a, by decreasing the distance between the donor and the acceptor, PL
11 decay kinetics of the NPLs significantly changes and becomes progressively faster, which
12 indicates that a new channel is opened up for relaxation of the carriers. This transition can be
13 attributed to FRET or charge transfer between the donor and the acceptor. However, in our solid
14 system, the surfaces of the NPLs are passivated by well-known nonaromatic ligand (oleic acid).
15 Additionally, the donor and the acceptor are separated with full-coverage from each other by the
16 ALD-grown spacer. Therefore, the charge transfer is ineffective in our hybrid structure, which is
17 also fully consistent with previous experimental studies.^{27,29,37}
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20 To analyze PL decay kinetics of the donor in the presence of the acceptor, PL decay curves were
21 fitted by multi-exponential functions, and their lifetime components and amplitude-weighted
22 contributions are summarized in Tables S1-S10. (See Supporting Information.) The amplitude-
23 average lifetime (τ_{av}) is plotted as a function of the Al₂O₃ thickness in Figure 2d.
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26 If the lifetime of the donor in the absence of acceptor is τ_D and the lifetime of the donor in the
27 presence of the acceptor is τ_{DA} , the total FRET rate (Γ_{FRET}) can be considered as $\Gamma_{FRET} =$
28 $1/\tau_{FRET} = 1/\tau_{DA} - 1/\tau_D$.^{28,38-40} Figure 2c shows Γ_{FRET} calculated from the experimental
29 measurements. Furthermore, we numerically computed Γ_{FRET} using an electromagnetic model
30 where the donor is represented as radiating dipoles in x , y , and z orientation inside each NPL in the
31 presence of lossy media modeled as a layer of MoS₂ placed between Al₂O₃ (used as the separating
32 film) and SiO₂ (taken as the substrate). The total theoretical FRET rate was obtained by the
33 arithmetic mean of FRET rates in the three orientations, given as the dash-dotted blue line in Figure
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3 2c. The experimental Γ_{FRET} was found to be 268.06 ns⁻¹ for the sample having 1 nm thickness. To
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5 the best of our knowledge, this attained Γ_{FRET} is the highest value that has been reported thus far
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7 in this kind of structure when compared to the previous studies.^{24,26-29} Consequently, as presented
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9 in Figure 2c, using a Al₂O₃ separating layer with thickness below 5 nm, FRET occurs in an ultrafast
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11 timescale corresponding to 3.73- 46.40 ps (see Figure 2b). These ultrafast FRET times overcome
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13 the Auger recombination process in NPLs, which is in the range of hundreds of picoseconds (150-
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15 500 ps) for CdSe NPLs,²¹ making these hybrid FRET structure potentially interesting for the
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17 purpose of reducing optical gain threshold in lasing systems and increasing charge extraction
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19 performance in multiexcitonic photovoltaics.
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24 Figure 2c depicts the numerical calculated FRET rate using COMSOL along with the experimental
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26 FRET rate extracted from TRF donor PL lifetimes. We see good agreement between the
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28 experimental observation and our classical electromagnetic model (see below for more details
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30 about the numerical calculations). In this figure, we observe that for distances larger than 20 nm,
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32 both the experimental and numerical FRET rates have an oscillating behavior. We attribute this
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34 oscillation to the electric field interference coming when the dipoles are placed on the top of
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36 dielectric surface. Figure 2d illustrates the FRET efficiency (η_{FRET}) of the samples, which are
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38 calculated by $\eta_{FRET} = 1 - (\tau_{DA}/\tau_D)$ as a function of the donor-acceptor distance (d). Here we
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40 find near-unity (99.88%) FRET efficiency for $d = 3.57$ and 5.07 nm. In addition, we further
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42 investigated the FRET rate distance dependency using the expression $\eta_{FRET} = 1/(1 + (\frac{d}{d_0})^n)$,
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44 where d_0 is the Förster radius and n is the parameter that indicates the FRET distance dependency.
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48 By fitting the experimental data numerically, we extract a Förster radius of $d_0 = 32.97 \pm 0.15$ nm
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50 and $n = 2$ order value (the numerical fitting for $n = 4$ which is the extracted order dependence for
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52 two dimensional acceptors, is given in Figure S5). $n = 2$ indicated that distance dependency of the
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interaction results from layer-to-layer dipole-dipole coupling.^{18,27,38–44} Consequently, the FRET distance dependency of $1/d^2$ leads to further extend the FRET radius to longer distances (> 10 nm) as compared to $1/d^4$ (the case of two dimensional structures) and $1/d^6$ (the case of zero dimension structures, *i.e.*, point to point).^{38,40} It is worth mentioning that the Förster radius ~ 32.97 nm is the longest that has been reported in literature thus far, to the best of our knowledge.

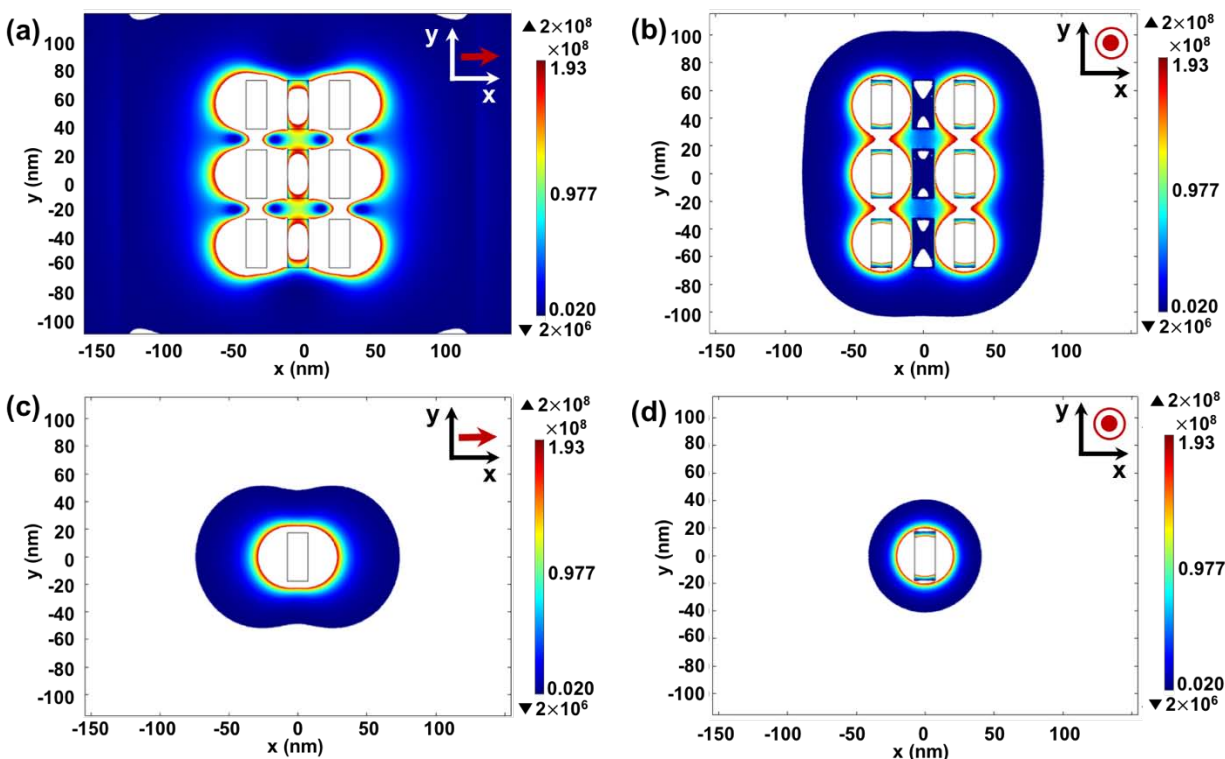


Figure 3. Electric field intensity (E^2) distribution at the NPL site: a) for several NPLs when the dipoles are oriented in the x axis and b) in the z axis, and c) for single NPL when the dipole is oriented in the x axis, and d) in the z axis.

To support the experimental results, we numerically computed the FRET rate. First, we assumed that an exciton dipole is placed inside several NPLs, and MoS_2 monolayer acts as exciton absorber medium (Figure 1a). Then, the FRET rate, Γ_{FRET} , for an exciton dipole with orientation α , is calculated by:^{45,46}

$$\Gamma_{FRET} = \frac{2}{\hbar} \frac{\text{Im}(\epsilon_{\text{MoS}_2}(\omega_{exc}))}{4\pi} \int_{\text{MoS}_2} \mathbf{E} \cdot \mathbf{E}^* dV \quad (1)$$

where $\varepsilon_{\text{MoS}_2}(\omega_{\text{exc}})$ is the MoS₂ dielectric function at the exciton frequency of the NPL and \mathbf{E} is the electric field induced by an oscillating exciton dipole $\mu e^{-i\omega_{\text{exc}}t}$. The integral was taken over the MoS₂ layer.

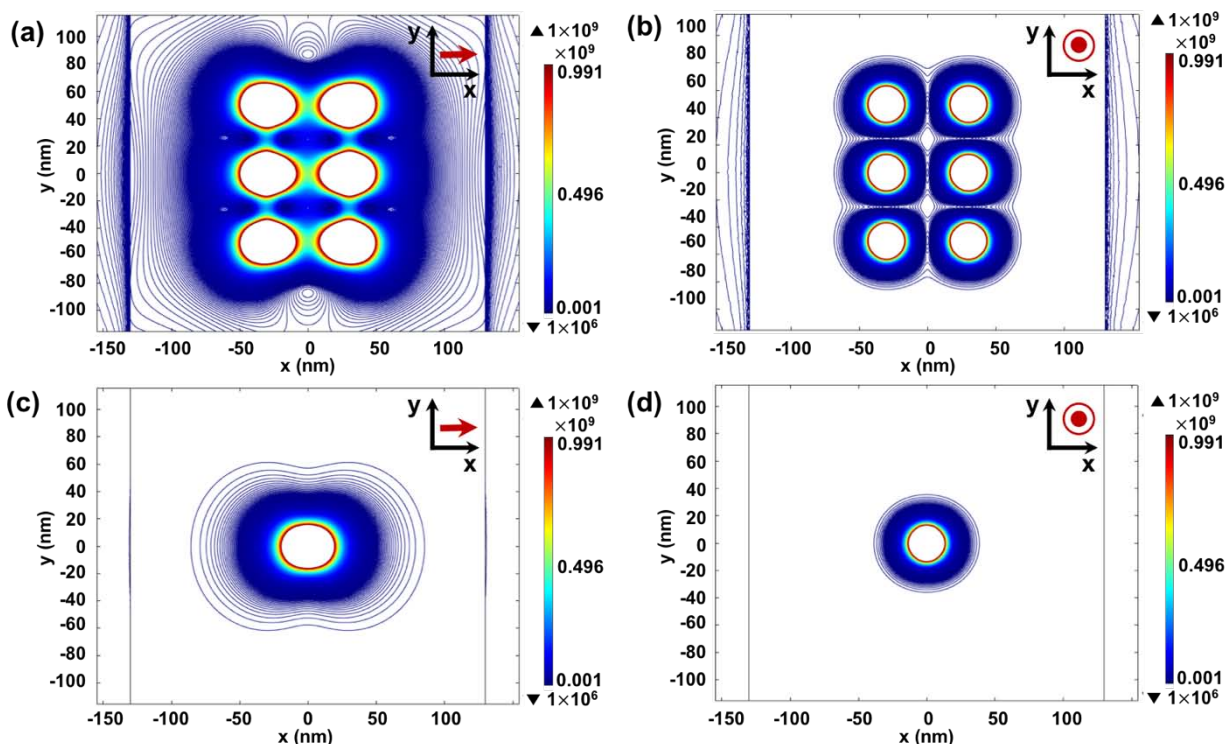


Figure 4. Electric field intensity (E^2) distribution at MoS₂ layer for 1 nm separation layer: a) for several NPLs when the dipoles are oriented in the x axis and b) in the z axis, and c) for single NPL when the dipole is oriented in the x axis and d) in the z axis.

The electric field was calculated using COMSOL, where the oscillating dipoles were placed inside the NPLs, and then the field induced subsequently in the MoS₂ layer was computed. The transfer rate, Γ_{FRET} , was calculated by using Eq 1 for different orientations of the dipole. Figure 3 shows the electric field intensity at the NPL site. As can be observed, E^2 is distributed along the layer of NPLs, which results from the delocalization of the dipoles. However, in the case of a single NPL, the field is mainly localized around the NPL.

Figure 4 shows the electric field intensity in the MoS₂ layer together with a 1 nm thick separation layer between the donor and the acceptor. It can be seen that the electric field for a x dipole is more delocalized as compared to a dipole oriented in the z axis. We attribute this behavior to the screening effect resulting from the dipole orientation. Figure 5 shows the electric field intensity distribution at MoS₂ layer when the dipole is oriented in the x and z axes using a 30 nm thick spacer. As can be observed from this figure, the electric field intensity is more distributed across

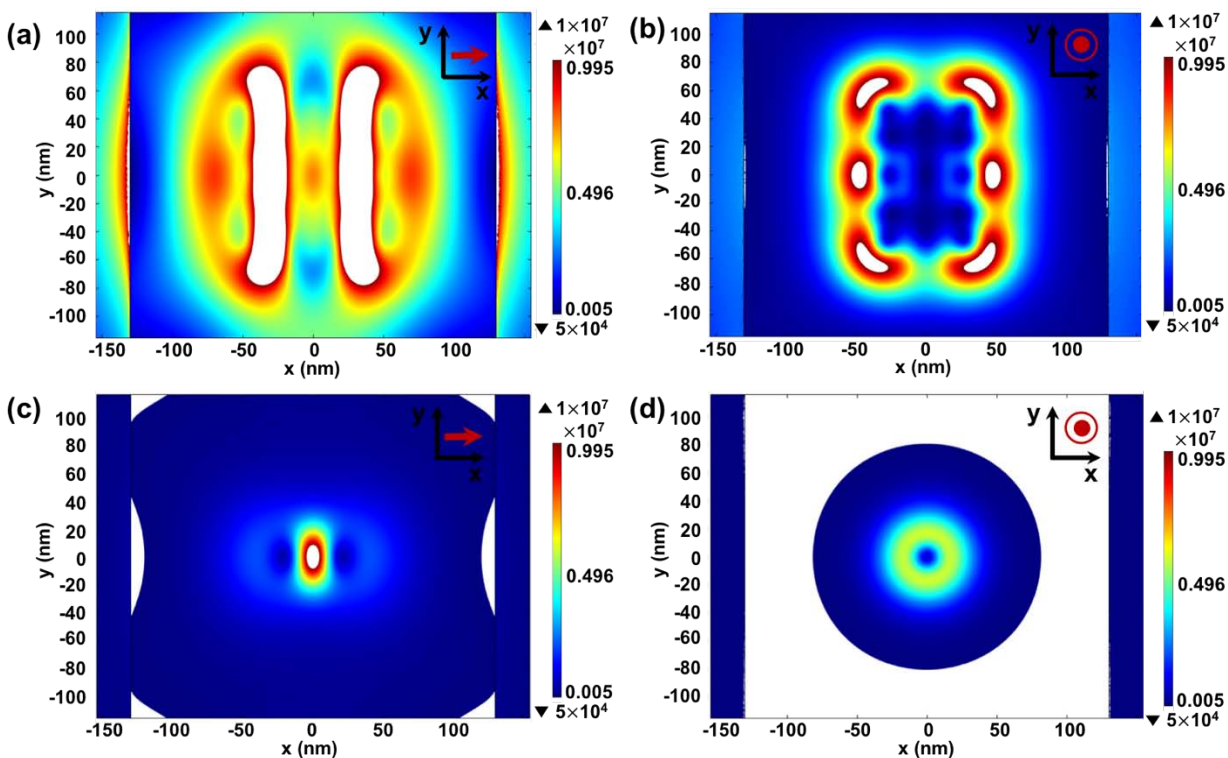


Figure 5. Electric field intensity (E^2) distribution at MoS₂ layer for a 30 nm thick separation layer: a) for several NPLs when the dipoles are oriented in the x axis and b) in the z axis, and c) for single NPL when the dipole is oriented in the x axis and d) in the z axis.

the MoS₂ in the case of several NPLs as compared to the single NPL where the electric field is more localized. Moreover, the electric field intensity is an order of magnitude larger for the x -oriented dipole than the z -oriented dipole, which emphasizes the anisotropic behavior of the CdSe based NPLs dipoles, as shown in the previous studies.^{47,48} Thus, we can imply that the main

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3 contribution for the FRET rates comes from the x - and y -oriented dipoles. Therefore, the exciton
4 delocalization along the NPLs and the electric field intensity distributed along the MoS₂ acceptor
5 layer support the FRET rate leading to the $1/d^2$ decay.^{18,27,38–44}
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8 9 10 **Conclusions**

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12 In conclusion, we have experimentally showed a Förster-type nonradiative energy transfer in the
13 hybrid structure of CdSe/CdS NPLs and MoS₂ monolayer thin film assembled together. We found
14 near-unity FRET efficiency reaching 99.88% with an ultrafast FRET rate of 268.06 ns⁻¹. Owing to
15 the layer-to-layer dipole coupling between the CdSe/CdS donor and the MoS₂ acceptor, the rate of
16 FRET follows the d^{-2} scaling. This dependency is supported with a full numerical electromagnetic
17 model, which shows that the d^{-2} behavior comes from the delocalization of the electric field
18 distribution across the layers of NPLs and MoS₂. In addition, the d^{-2} dependency further extends
19 the Förster radius to a value of ~ 33 nm, which is the longest reported to date, to the best of our
20 knowledge.
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33 **Methods**

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35 ***Time-resolved fluorescence (TRF) spectroscopy.*** TRF spectroscopy was carried out using *via*
36 Pico Quant FluoTime 200 spectrometer. The NPLs were excited by a picosecond pulsed laser at
37 the wavelength of 375 nm at low level intensity (≈ 10 nJ/cm²). Thus, the number of photogenerated
38 excitons was set much smaller than one photon per NPL on the average ($\langle N \rangle \ll 1$). The
39 fluorescence decay curves were collected by TimeHarp time-correlated single-photon counting
40 (TCSPC) unit. Then, we fitted the fluorescence decay curves to multiexponential decays through
41 the FluoFit software by using the reconvolution mode of fitting to account for the instrument
42 response function (IRF) of the excitation pulse.
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Supporting Information

Synthesis and characterization of 4 monolayer CdSe/CdS core/crown nanoplatelets, the growth of MoS₂ flakes *via* chemical vapor deposition, sample characterization by atomic force microscopy. Time-resolved fluorescence decay curves from different positions of the samples along with their analyses. The numerical fitting of Förster resonance energy transfer efficiency.

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Conflict of Interest

The authors declare no financial competing financial interest.

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